

REGIOSELECTIVE AND [C,C]CONNECTIVE ROUTES TO OXETANES AND TETRAHYDROFURANES

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α -Selenoalkyllithiums already used for the regioselective synthesis of epoxides¹ are found valuable building blocks for the synthesis of higher homologues namely oxetanes and tetrahydrofuranes. The general synthetic strategy used is schematized below and represent a formal homologization reaction of epoxides and oxetanes by carbene insertion.

SCHEME I



The synthetic approach requires the synthesis of γ and δ -hydroxyselenides 3 (Scheme II) from α -selenoalkyllithiums and epoxides² or oxetanes (THF, -78° to $+20^\circ$, method A, Table). β -Hydroxyselenides were found as side products beside the desired γ -hydroxyselenides when methyl or phenylselenomethylolithiums are reacted. They can be accounted from the reaction, on epoxides of lithioselenolates probably formed by decomposition of α -selenomethylolithiums. However, their formation can be prevented if HMPT is present in the medium (2 in THF/HMPT, 1 eq. added at -78° on 1 then heated up to $+20^\circ$, Method B).

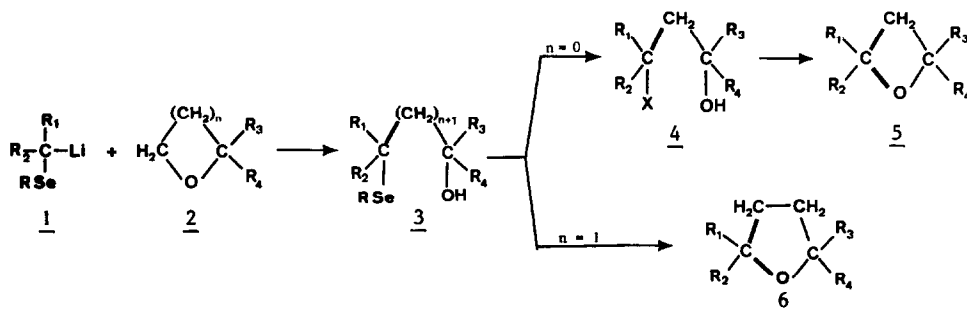
δ -Hydroxyselenides (3, $n=1$) are directly cyclized in good yield to tetrahydrofuranes on reaction with bromine³ [$\text{Br}_2/\text{NEt}_3/\text{CH}_2\text{Cl}_2$, 20° , 0.5 hr (method C) or $\text{Br}_2/\text{ethanol-water}$ (method D)]. γ -Hydroxyselenides (3, $n=0$) however do not cyclize to oxetanes under similar (methods C, D) and related reaction conditions³ [$\text{CH}_3\text{I}/\text{NaI}/\text{DMF}, \text{CaCO}_3$, 80° , 1hr (method E)]. The reactions stop at the stage of γ -halogenoalcohols³ which are produced in good yield^{3b} (table) and which require a basic media to cyclize to oxetanes 5.

The choice of the right basic system is crucial for the success of the cyclization which is directly related to the number and the nature of the substituents branched on the carbon bearing the halogen atom.

tBuOK/DMSO (20° , method F) is the most valuable system for the cyclization⁴ of γ -halogenohydrins bearing the halogen atom at the terminus ($\text{R}_1, \text{R}_2 = \text{H}$, Table, entries 1 to 3). $\text{CH}_3\text{MgBr}/\text{HMPT}$ (20° to 80° , method G)⁵ is the only one among a huge variety of basic systems used^{4,6} to allow the desired synthesis when a methyl group is attached at the terminus ($\text{R}_1=\text{H}$, $\text{R}_2=\text{CH}_3$, Table, entries 4 to 5).

Under these conditions, higher homologues produce low yields of oxetanes ($R_1=H$, $R_2=CH_3$ 16%; $R_1=H$, $R_2=C_6H_{13}$ ~3%) and homoallyl alcohols are instead formed² (respectively 51% and 70% yields). Work is actually in progress to generalize the oxetane synthesis from epoxides.

SCHEME II



TABLE

entry	R_1	R_2	n	R_3	R_4	3 yield %		4 yield %		5 or 6 yield %		
						R	(method)	X	(method)	5 or 6	method	
1	H	H	0	H	C_6H_{13}	C_6H_5	46(A) ^a	80(B) ^b	I	88(E)	5	70(F)
2	H	H	0	H	C_6H_{13}	CH_3	25(A) ^c	73(B) ^d	I	71(E)	5	70(F)
3	H	H	0	CH_3	C_6H_{13}	C_6H_5		75(B)	I	60(E)	5	70(F)
4	CH_3	H	0	H	C_6H_{13}	C_6H_5	70(A) ^e		Br	67(D) ^e	5	60(G) ^e
5	CH_3	H	0	CH_3	C_6H_{13}	C_6H_5	50(A) ^e	60(B)	Br	55(D) ^e	5	65(G) ^e
6	C_2H_5	H	0	H	C_6H_{13}	CH_3	75(A) ^e		Br	74(D) ^e	5	16(G) ^e
7	C_6H_{13}	H	0	H	CH_3	C_6H_5	84(A) ^e		Br	74(D) ^c	5	3(G)
8	C_6H_{13}	H	1	H	H	CH_3	82(A)				6	77(C)(D)
9	C_6H_{13}	H	1	H	H	C_6H_5	75(A)				6	70(C)(D)
10	C_6H_{13}	H	1	CH_3	H	CH_3	42(A) ^e				6	84(C) ^e
11	C_6H_{13}	CH_3	1	H	H	H	64(A)				6	82(C)

1-seleno 2-hydroxy octane is obtained in a) 26% yield, b) 3% yield, c) 31% yield, d) 10% yield, e) these derivatives are obtained as a mixture of stereoisomers.

References

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3. a) For a report on reaction of bromine with selenides: M. Sevrin, W. Dumont, L. Hevesi and A. Krief, *Tet. Lett.*, 2647 (1976)
b) We are not able at present to achieve the desired transformations (method A or B) when the selenyl moiety is attached to a carbon atom fully substituted.
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